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Discovery of trisubstituted cyclohexanes as potent CC chemokine receptor 2 (CCR2) antagonists

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ABSTRACT

A series of trisubstituted cyclohexanes was designed, synthesized and evaluated as CC chemokine receptor 2 (CCR2) antagonists. This led to the identification of two distinct substitution patterns about the cyclohexane ring as potent and selective CCR2 antagonists. Compound **36** exhibited excellent binding (CCR2 IC_{50} = 2.4 nM) and functional antagonism (calcium flux IC_{50} = 2.0 nM and chemotaxis IC_{50} = 5.1 nM).

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Chemokines are endogenous proteins typically produced in small amounts to control the activation and migration of leukocytes.¹ However, in many autoimmune and inflammatory conditions chemokines are overexpressed, and hence continue to recruit inflammatory cells into the tissues and joints.¹ We have been interested in monocyte chemoattractant protein-1 (MCP-1 or CCL2), a CC chemokine,² which elicits a functional response by binding to its G-protein coupled receptor (GPCR), CC chemokine receptor 2 (CCR2).³ With high levels of MCP-1 and CCR2 in many diseases, the pair is thought to play a role in rheumatoid arthritis,⁴ atherosclerosis,⁵ multiple sclerosis⁶ and insulin resistance.⁷ As a result, there has been significant interest in the design and synthesis of CCR2 antagonists as potential therapeutic agents.⁸ In this communication, we describe our discovery and initial structure–activity relationship (SAR) study of trisubstituted cyclohexanes as novel CCR2 antagonists.

Recently, we described a series of disubstituted cyclohexanes and piperidines (Fig. 1, representative compounds **1** and **2**, respectively) as selective CCR2 antagonists.⁹ Some piperidines were as much as 180-fold more active than the corresponding disubstituted cyclohexanes, and the piperidine nitrogen was suspected of contacting the glutamic acid residue at position 291 within the CCR2 receptor.^{9b} In an attempt to optimize and explore this, we began an investigation of trisubstituted cyclohexanes **3** with an amino group in the 4- or 5-position, as shown in Figure 1.

The trisubstituted cyclohexanes **12–16** were synthesized in racemic form as shown in Scheme 1. Starting from the known racemic *cis*-epoxide **4**,¹⁰ the epoxide was opened with sodium azide to give **5**. The azide was treated with triphenylphosphine and water to afford an amine, which was converted to carbamate **6**. The benzyl and trifluoroacetamide groups were removed and were replaced with a Cbz carbamate to furnish **7**. Inversion of the secondary alcohol to azide **8** was accomplished via a Mitsunobu reaction. The azide was converted to amine **9**, which was capped as benzamide **10**. The glycineamide was installed in one operation to give **12** or **13**, depending on the R¹ group of **11**. Compound **13** was deprotected with acid to yield **14**, which was converted to urea **15** via treatment with methyl isocyanate. The *iso*-propyl amine **16** was produced from **14** by a reductive amination.

As shown in Scheme 2, the synthesis of the racemic *cis*-targets **24–28** utilized an analogous sequence as above, however it started with racemic *trans*-epoxide **17**.¹⁰

Targets with substitution at the 4-position were realized by direct glycineamide coupling to racemic intermediate **9** (see Scheme 3). The resulting *bis*-amide **29** was converted to target **30** via carbamate removal and benzamide coupling. In addition, compound **30** was deprotected to **31**, which in turn was converted to **32** via the previously described reductive amination.

Utilizing the above strategy, racemic intermediate **22** was converted to benzamide **34** (see Scheme 4). Deprotection of **34** and glycineamide installation gave either **35** or **37**, depending on the R¹ of **11**. Compounds **36** and **38** were then produced via the deprotection of **35** and **37**, respectively.

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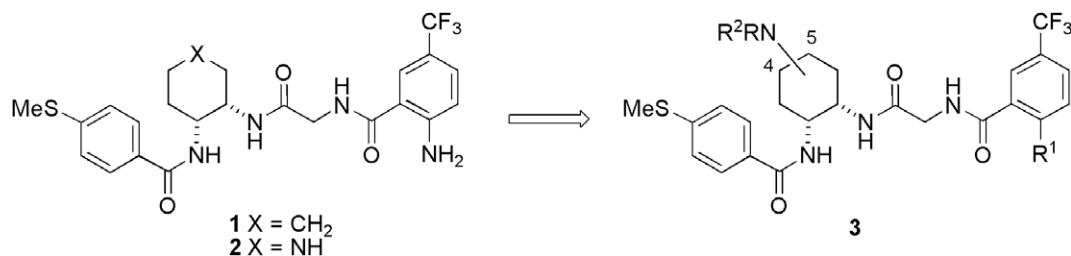
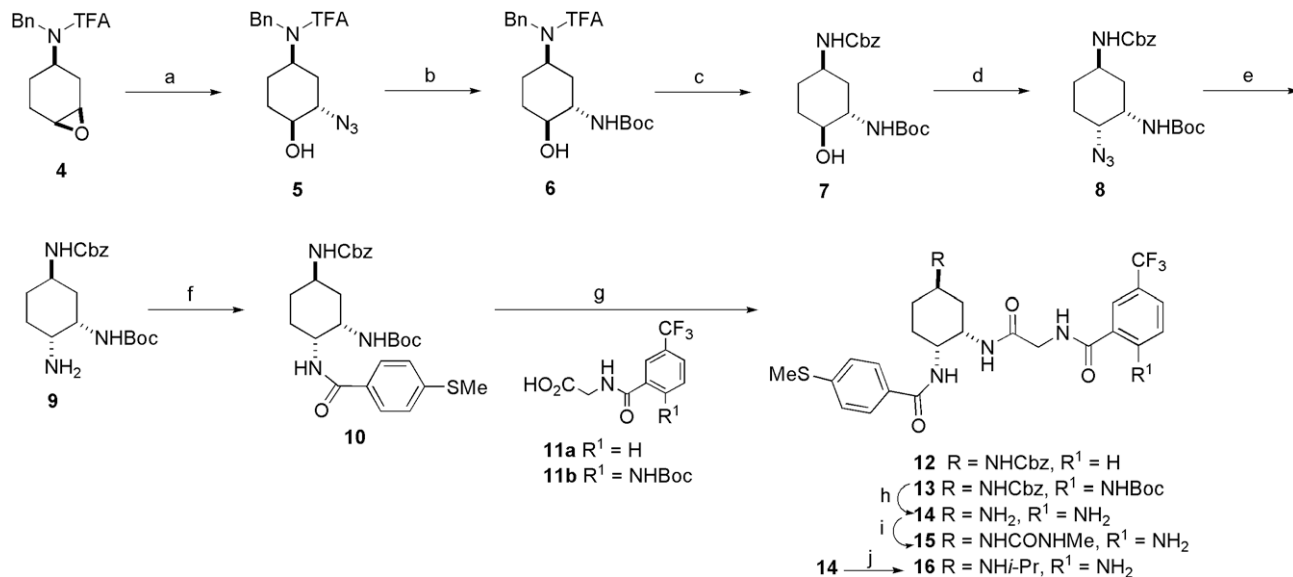
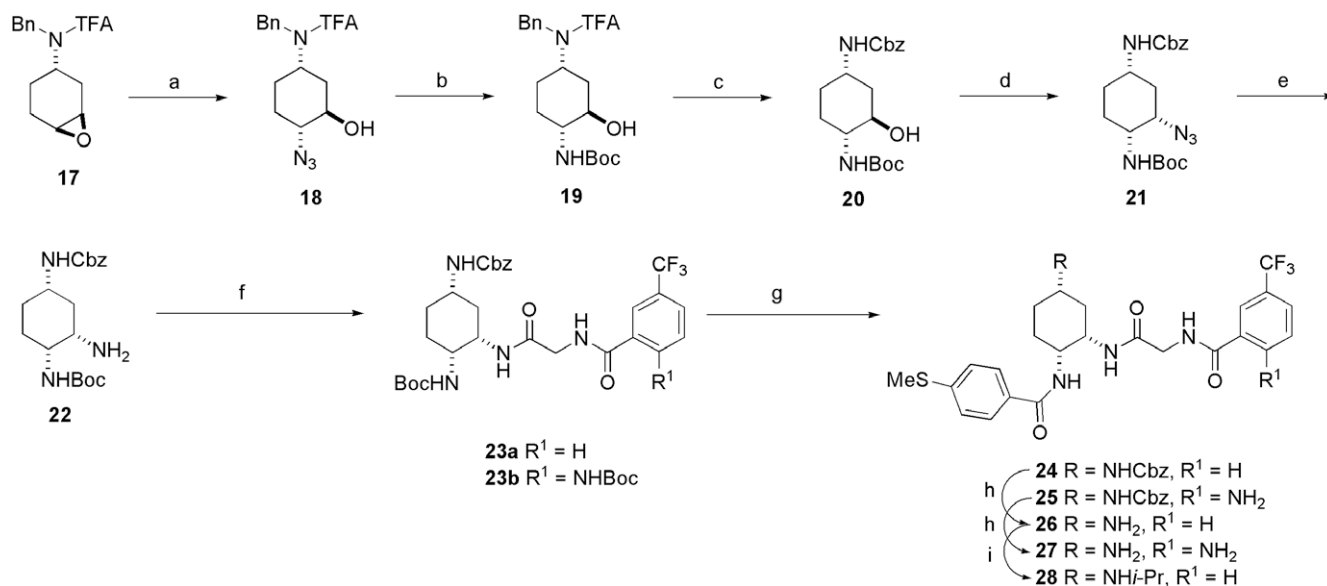


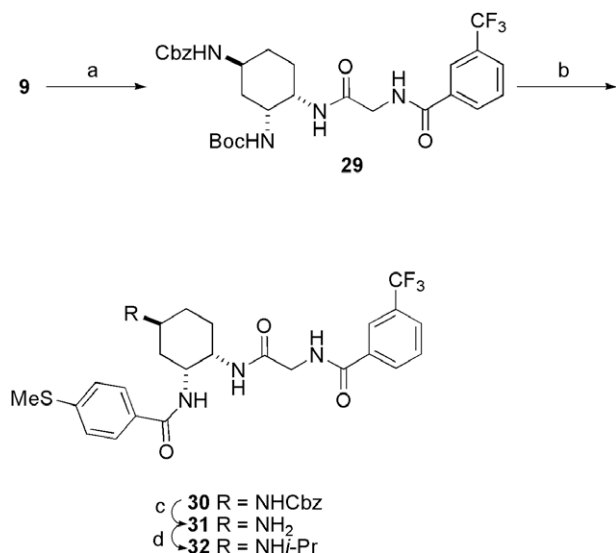
Figure 1. Proposed exploration of trisubstituted cyclohexanes **3** as CCR2 antagonists.



Scheme 1. Reagents and conditions: (a) NaN₃, MeOH/H₂O, Δ, 74%; (b) i-PPh₃, H₂O; ii-(Boc)₂O, NaHCO₃, THF/H₂O, 82% for two steps; (c) i-KOH, H₂O, MeOH; ii-H₂(50 psi), 5% Pd/C, MeOH; iii-CbzCl, NaHCO₃, THF/H₂O, quant; (d) HN₃, DEAD, PPh₃, THF/PhH, 5 °C, 93%; (e) PPh₃, H₂O, THF, 79%; (f) 4-(thiomethyl)benzoic acid, (i-Pr)₂NEt, HATU, DMF, 94%; (g) i-TFA, CH₂Cl₂; ii-**11a** or **11b**, (i-Pr)₂NEt, HATU, DMF, 87% for **13**; (h) 38% HBr/HOAc, 69%; (i) MeNCO, (i-Pr)₂NEt, CH₂Cl₂, 27%; (j) acetone, HC(OMe)₃, NaBH(OAc)₃, 37%.



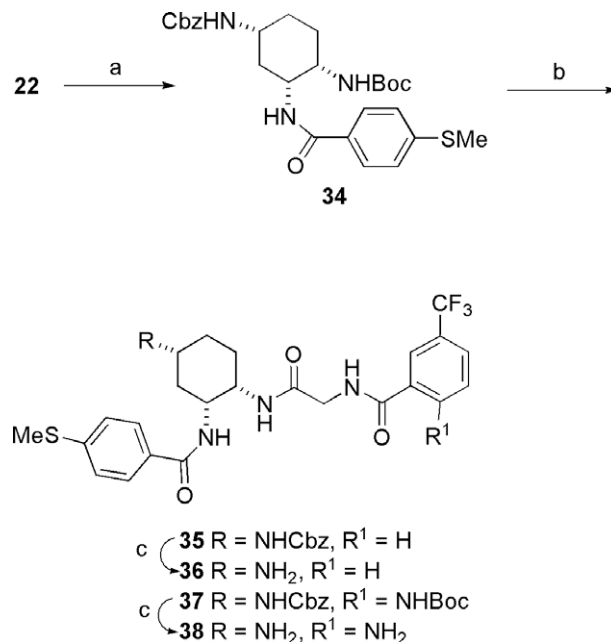
Scheme 2. Reagents and conditions: (a) NaN₃, MeOH/H₂O, Δ, 57%; (b) i-PPh₃, H₂O; ii-(Boc)₂O, NaHCO₃, THF/H₂O, 80% for two steps; (c) i-KOH, H₂O, MeOH; ii-H₂(50 psi), 5% Pd/C, MeOH; iii-CbzCl, NaHCO₃, THF/H₂O, 88%; (d) HN₃, DEAD, PPh₃, THF/PhH, 5 °C, 82%; (e) PPh₃, H₂O, THF, 90%; (f) **11a** or **11b**, (i-Pr)₂NEt, HATU, DMF, 87%; (g) i-TFA, CH₂Cl₂; ii-4-(thiomethyl)benzoic acid, (i-Pr)₂NEt, HATU, DMF, 62% for **23b**; (h) 38% HBr/HOAc; (i) acetone, HC(OMe)₃, NaBH(OAc)₃, 68%.



Scheme 3. Reagents and conditions: (a) **11a**, (*i*-Pr)₂NEt, HATU, DMF, 86%; (b) *i*-TFA, CH₂Cl₂, ii—4-(thiomethyl)benzoic acid, (*i*-Pr)₂NEt, HATU, DMF, 38%; (c) HBr/HOAc, 90%; (d) acetone, HC(OMe)₃, NaBH(OAc)₃, NaBH(OAc)₃, 81%.

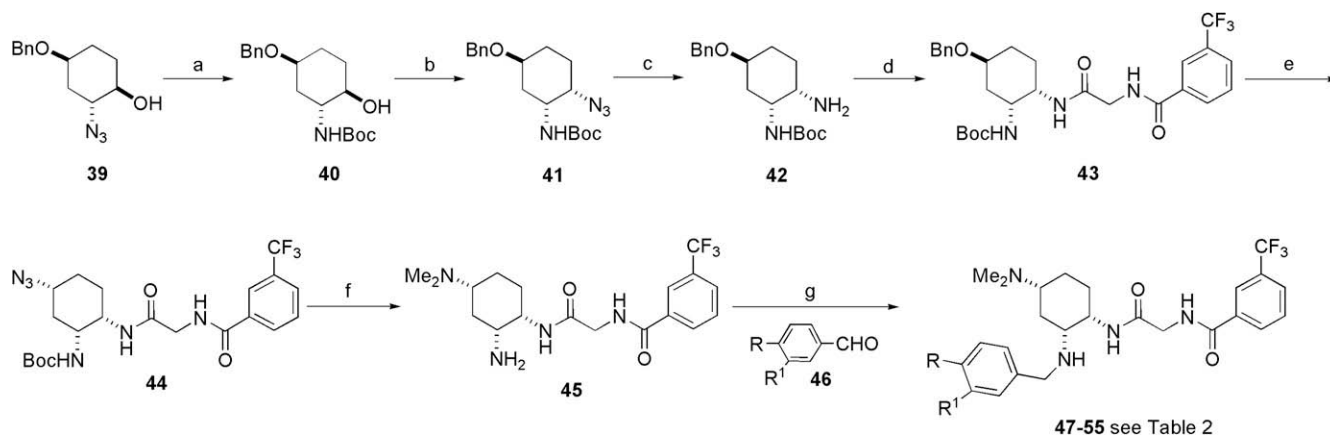
Benzyl amine targets were synthesized according to the sequence illustrated in Scheme 5. Starting with the known azido alcohol **39**,¹¹ reduction and carbamate formation were accomplished in one step to give **40**. The alcohol was converted to a mesylate prior to an azide displacement to afford **41**. Reduction to amine **42** was followed by glycinamide formation to provide **43**. After hydrogenolysis of the benzyl group, another alcohol to azide sequence gave **44**. The azide was reduced and was dimethylated before the final carbamate removal revealed **45**. The final targets **47–55** were realized via reductive amination of **45** with the appropriate aldehyde.

The trisubstituted cyclohexanes were evaluated in vitro, using peripheral blood mononuclear cells (PBMCs) in a radiolabeled MCP-1 displacement assay.¹² We were interested in selective CCR2 antagonists and initially used a CCR3 binding assay¹³ for a quick assessment of selectivity. Compounds with CCR2 binding IC₅₀s <20 nM were also evaluated in two functional assays: a calcium flux assay^{12,14} and a chemotaxis assay.¹² As shown in Table 1, the first compounds tested in the CCR2 binding assay, trisubstituted analogs **12** and **13**, did manage some affinity for CCR2, even though the carbamate groups were still in place. Removal of

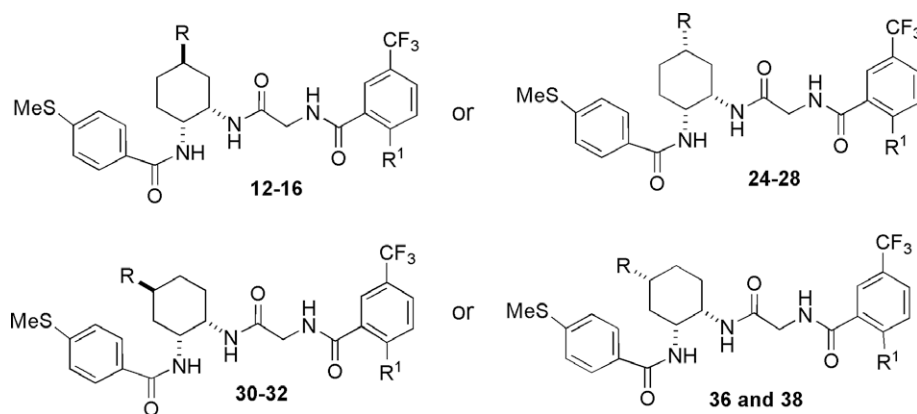


Scheme 4. Reagents and conditions: (a) 4-(thiomethyl)benzoic acid, (*i*-Pr)₂NEt, HATU, DMF, 98%; (b) *i*-TFA, CH₂Cl₂; ii—**11a** or **11b**, (*i*-Pr)₂NEt, HATU, DMF, 85%; (c) 38% HBr/HOAc, 48%.

the carbamate group from **13** resulted in a 4-fold increase in CCR2 affinity for amine **14**. Two beneficial substitutions were the urea **15** and the *iso*-propyl **16**. Compound **16** was significant as it displayed a 41-fold increase and a 7-fold increase in affinity for CCR2 as compared to the previously described disubstituted cyclohexane **1^{9a}** and piperidine **2^{9b}**, respectively. Compound **16** also demonstrated excellent activity in the calcium flux assay while remaining selective against CCR3. In general, inversion of the 5-position (compounds **24–28**) did not improve affinity for CCR2, although the *iso*-propyl amine **28** displayed a CCR2 binding IC₅₀ = 135.7 nM. The *trans*-compounds in the 4-position (**30–32**) were also lacking in CCR2 activity with the primary amine **31** (CCR2 IC₅₀ = 227.5 nM) representing the best in this class. However, inversion at the 4-position gave the *cis*-compound **36**, which had surprising activity in binding (CCR2 IC₅₀ = 2.4 nM), calcium (CCR2 IC₅₀ = 2.0 nM), and chemotaxis (CCR2 CTX IC₅₀ = 5.1 nM). The 2-amino compound **38** did serve as a direct comparison to



Scheme 5. Reagents and conditions: (a) H₂, Pd(OH)₂, (Boc)₂O, EtOAc, 89%; (b) *i*-MsCl, Et₃N, CH₂Cl₂, ii—NaN₃, CH₂Cl₂, DMSO, Δ, 65%; (c) H₂, 10% Pd/C, MeOH; (d) **11a**, BOP, NMM, DMF, 95%; (e) *i*-H₂ (50 psi), Pd(OH)₂, MeOH; ii—MsCl, Et₃N, CH₂Cl₂; iii—NaN₃, DMSO, Δ, 50%; (f) *i*-H₂, 10% Pd/C, MeOH; ii—37% HCHO, NaBH₃CN, MeOH, iii—TFA; (g) NaBH(OAc)₃, THF.

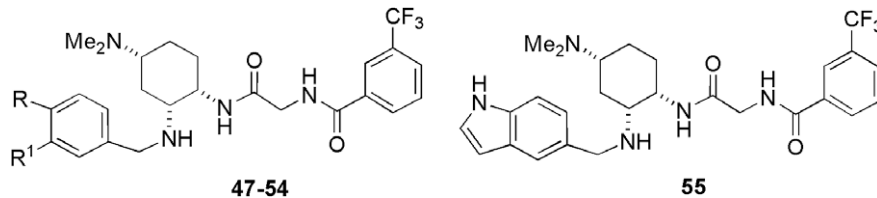
Table 1Evaluation of trisubstituted cyclohexane derivatives^a

Compound	R	R ¹	IC ₅₀ ^b (nM)			CCR3 binding %Inh at 10 μM ^c
			CCR2 binding	Ca ²⁺ flux	Chemotaxis	
1	See Figure 1		155 ± 55.2 (2)	NT	NT	NT
2	See Figure 1		28 ± 7.1 (2)	75 ± 0.7 (2)	109 ± 58.0 (2)	19.7
12	NHCbz	H	1605.0 ± 572.0 (2)	NT	NT	NT
13	NHCbz	NHBoc	600.0 ± 247.5 (2)	NT	NT	NT
14	NH ₂	NH ₂	167.5 ± 74.2 (2)	NT	NT	NT
15	NHCONHMe	NH ₂	112.0 ± 14.1 (2)	NT	NT	NT
16	NHi-Pr	NH ₂	3.8 ± 1.1 (2)	2.0 (1)	NT	27.2 ± 4.9 (2)
24	NHCbz	H	0% at 1 μM	NT	NT	NT
25	NHCbz	NH ₂	3.1% at 1 μM	NT	NT	NT
26	NH ₂	H	13% at 1 μM	NT	NT	NT
27	NH ₂	NH ₂	28.1% at 1 μM	NT	NT	NT
28	NHi-Pr	H	135.7 ± 8.5 (4)	NT	NT	NT
30	NHCbz	H	14% at 1 μM	NT	NT	NT
31	NH ₂	H	227.5 ± 34.6 (2)	NT	NT	NT
32	NHi-Pr	H	31.4% at 1 μM	NT	NT	NT
36	NH ₂	H	2.4 ± 0.9 (3)	2.0 (1)	5.1 ± 3.8 (2)	7.0 ± 6.2 (2)
38	NH ₂	NH ₂	4.8 ± 2.5 (2)	6.0 (1)	13.5 ± 8.2 (3)	NT

^a Compounds are racemic, one enantiomer is displayed for illustrative purposes.^b IC₅₀ values (n) are displayed as mean ± SD (n = 2) and mean ± SEM (n > 2).^c CCR3% inhibition are n = 1, unless otherwise noted; NT, not tested.

cyclohexane **1** and piperidine **2**, where compound **38** displayed 32-fold and 6-fold more CCR2 affinity than the disubstituted compounds, respectively. Significantly, the 2-amino-benzamide compound **38** did not improve CCR2 functional antagonism as

compared to compound **36**. In this regard, compound **36** is unusual, as previous studies^{9a,15} have shown a reliance on the 2-amino benzamide group (of the glycinamide) for potent functional activity.

Table 2Evaluation of trisubstituted cyclohexane derivatives^a

Compound	R	R ¹	IC ₅₀ ^b (nM)		
			CCR2 binding	Ca ²⁺ flux	Chemotaxis
47	SMe	H	2.4 ± 0.6 (2)	6.0 (1)	22.5 ± 7.8 (2)
48	Me	H	9.9 ± 8.5 (2)	5.5 ± 0.7 (2)	12.6 ± 4.8 (3)
49	<i>i</i> -Pr	H	16.5 ± 0.7 (2)	19 (1)	NT
50	OMe	H	12.5 ± 0.7 (2)	8.0 (1)	41.4 ± 26.6 (4)
51	Cl	H	16.0 ± 0.0 (2)	7.0 (1)	NT
52	Cl	Cl	61.5 ± 4.9 (2)	NT	NT
53	H	CF ₃	32.5 ± 17.7 (2)	27 (1)	NT
54	NHAc	H	7.7 ± 3.0 (2)	NT	17.4 ± 4.5 (6)
55	See structure		3.1 ± 1.1 (2)	8.0 (1)	23.0 ± 17.0 (2)

^a Compounds are racemic, one enantiomer is displayed for illustrative purposes.^b IC₅₀ values (n) are displayed as mean ± SD (n = 2) and mean ± SEM (n > 2). NT, not tested.

As a result of this activity, we investigated benzyl amines, as shown in Table 2. The 4-thiomethyl of **47** remained one of the better substituents, although 4-methyl **48** and 4-*N*-acetamide **54** only lost 4-fold in affinity for CCR2 as compared to **47**. Other substituents at the 4-position (**49**, **50**, **51**) lost additional affinity as compared to **47**. Substitution as 3,4-dichloro (**52**) or 3-trifluoromethyl (**53**) was not favorable; however the indole **55** was approximately equipotent to **47** in both binding and functional activity.

In summary, we have described the use of selective epoxide openings as a unified approach toward the synthesis of four new trisubstituted scaffolds as novel CCR2 antagonists. All four substitution patterns afforded CCR2 antagonists with binding $IC_{50} < 250$ nM. A substantial enhancement in CCR2 binding affinity was observed for two of the trisubstituted scaffolds (see compounds **16** and **36**) as compared to the disubstituted cyclohexane antagonists reported earlier. Significantly, compound **36** displayed potent functional antagonism of CCR2 without resorting to the large lipophilic groups needed in previously described studies.

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